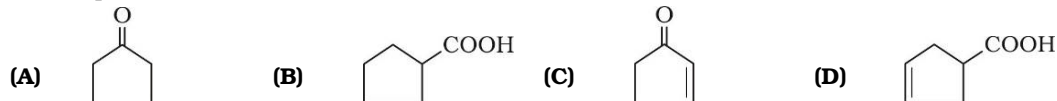
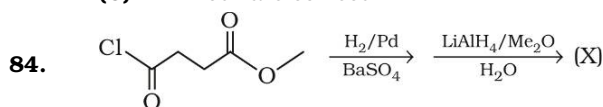


The compound B is :

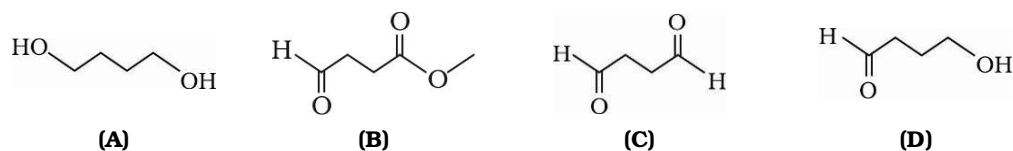


83. Peroxyacetic acid ($\text{CH}_3\text{CO}_3\text{H}$) is a weaker acid than acetic acid ($\text{CH}_3\text{CO}_2\text{H}$) since :

- (A) negative charge in CH_3COO^- can't be delocalised into the carbonyl group
(B) CH_3 group in $\text{CH}_3\text{CO}_3\text{H}$ shows +I effect
(C) Both are correct (D) None is correct



The structure of product (X) :



85. Give the structure of the expected product of the following reaction.

